



Introduction

- Chemical food preservatives → prevent or delay the spoilage of foods.
- The use of food preservatives is regulated by the Food and Drug Administration (FDA).衛生福利部食品藥物管理署
- The chemical preservatives listed in Table 13-1 are generally recognized as safe (GRAS).







Benzoic Acid and the Parabens

Mode of action:

- inhibit the cellular uptake of substrate molecules.
- The undissociated form is essential to the antimicrobial activity of benzoate as well as for other lipophilics such as sorbate and propionate.



Sorbic acid (己二烯酸)

- Sorbic acid is employed as a food preservative, usually as the calcium, sodium, or potassium salt. → are permissible in foods ≤ 0.2%.
- More effective in acid foods → used as fungal inhibitors → works best ≤ pH
 6.0
- more effective than sodium benzoate between pH 4.0 and 6.0



Sorbic acid

- pK = 4.8
- primarily effective against molds and yeasts
- also effective against Staphylococcus aureus, salmonellae, coliforms, psychrotrophic (嗜冷的) spoilage bacteria (especially the pseudomonads), and Vibrio parahaemolyticus



Sorbic acid

- The resistance of the lactic acid bacteria to sorbate→use as a fungistat in products that undergo lactic fermentations.
- The widest use of sorbates is as fungistats (抑真菌劑) in products such as cheeses, bakery products, fruit juices, beverages, salad dressings, and the like.



Sorbic acid

• use in **meat products** in combination with **nitrites**

- no significant differences are found in the organoleptic qualities or in botulinal protection.
 - 120 ppm NaNO2 without sorbate
 - 40 ppm NaNO2 and 0.26% potassium sorbate (was proposed by the U.S. Department of Agriculture (USDA) in 1978 but postponed in 1979. ←"chemical"-like flavors and producing prickly (刺痛的) mouth sensations



- ◆ Involves the proton motive force (PMF 質子驅動 力).
- Hydrogen ions (protons) and hydroxyl ions are separated by the cytoplasmic membrane, hydrogen ions (outside the cell) giving rise to acidic pH and hydroxyl ions (inside the cell) giving rise to pH near neutrality.



The antimicrobial mechanism of **lipophilic acids** (sorbate, benzoate, and propionate)

- The membrane gradient represents electrochemical potential that the cell employs in the active transport of some compounds such as amino acids.
- After these weak lipophilic acids diffusing across the membrane, the undissociated molecule ionizes inside the cell and lowers intracellular pH. → a weakening of the transmembrane gradient such that amino acid transport is affected adversely



Sorbic acid

• With respect to safety, **sorbic acid is metabolized in the body to CO₂ and H₂O** in the same manner as **fatty acids** normally found in foods.



Propionates (丙酸)

- This acid and its calcium and sodium salts are permitted in breads, cakes, certain cheese, and other foods, primarily as a mold inhibitor.
- permissible in foods $\leq 0.32\%$
- The pK is 4.87 → Active in low-acid foods (pH 4.6-6.8)



Sulfur Dioxide and Sulfite (二氧化硫及亞硫酸鹽)

- Sulfur dioxide (SO₂) and the sodium and potassium salts of sulfite (=SO₃,亞硫酸鹽), bisulfite (=HSO₃,重亞硫酸鹽), and metabisulfite (=S₂O₅,焦亞硫酸鹽) all act similarly.→in foods 200-300 ppm
- Sulfur dioxide is used in its gaseous or liquid form or salts on dried fruits, in lemon juice, molasses, wines, fruit juices, and others.
- It is **not permitted in meats** or other foods recognizable as **sources of thiamine** (Vitamin B1).



Sulfur Dioxide and Sulfite

- The sulfites react with various food constituents including **nucleotides**, **sugars**, **disulfide bonds**, and others.
- Also used as an **antioxidant**.
- SO₂ is **bacteriostatic** against *Acetobacter* spp. and the lactic acid bacteria at low pH, concentrations of 100-200 ppm being effective in fruit juices and beverages. It is **bactericidal at higher concentrations**.
- SO₂ also show **inhibition on spores of** *Clostridium botulinum* and **on the growth of salmonellae and other Enterobacteriaceae.**



Sulfur Dioxide and Sulfite

- Molds such as *Botrytis* can be controlled on grapes by periodic gassing with SO₂ and bisulfite can be used to destroy aflatoxins (黃麴毒素).
 - Both aflatoxins B1 and B2 can be reduced in corn.



Sulfur Dioxide and Sulfite

- The actual mechanism of action of SO2 is not known.
 - One suggestion is that the **undissociated sulfurous acid or molecular SO2** is responsible for the antimicrobial activity (Its greater effectiveness at low pH tends to support this).
 - The other suggestion is that the antimicrobial action is due to the strong reducing power that allows these compounds to reduce oxygen tension to a point below that at which aerobic organisms can grow or by direct action on some enzyme system.



Sulfur Dioxide and Sulfite

- SO₂ is also thought to be an enzyme poison, inhibiting growth of microorganisms by inhibiting essential enzymes.
 - Its **use in the drying of foods to inhibit enzymatic browning** is based on this assumption.
 - Because the sulfites are known to act on disulfide bonds, it may be presumed that certain essential enzymes are affected.



Nitrites and Nitrates (亞硝酸及硝酸)

- Sodium nitrate (NaNO3) and sodium nitrite (NaNO2) are used in curing formulas for meats because they
- stabilize red meat color,
- inhibit some spoilage and food poisoning organisms,
- contribute to **flavor development**.



Nitrites and Nitrates

- Many bacteria can utilize nitrate as an electron acceptor to produce nitrite.
- The nitrite ion is more important than the nitrate in preserved meats.
- The nitrite ion is highly reactive → serving as both a reducing and an oxidizing agent.
- In an acid environment, nitrite ion can be reduced to yield nitric oxide (NO)
 →important for color fixation in cured meats.











Nitrites and Nitrates

- The antibacterial effect of NO₂ increases as pH is lowered within the acid range ← increase in the undissociated HNO₂.
- The cooked cured meat pigment is dinitrosyl ferrohemochrome (DNFH). It forms when globin in nitrosomyoglobin is replaced with a second NO group.



Nitrites and Nitrates

Organisms Affected

- Although the greatest concern relative to nitrite inhibition is *Clostridium botulinum*, the compound is antimicrobial for other organisms.
 - Against other clostridia
 - Against *Staphylococcus aureus* at high concentrations.
- It is **ineffective against Enterobacteriaceae** (including the salmonellae) and the **lactic acid bacteria**.



The Perigo Factor (皮瑞果因子)

- Produced from heating of the culture medium with nitrite → 10 times more inhibitory to botulism than nitrite alone
- Heating to at least 100°C is necessary for its development, although some activity develops in meats when heated to as low as 70°C.
- Questionable in cured meats



Nitrites and Nitrates

- The antibotulinal activity of nitrite in cured meats is of greater public health importance than the facts of color and flavor development.
- For color and flavor development, initial nitrite levels as low as 15-50 ppm are enough for various meat products.
- The antibotulinal effect requires at least 120 ppm for bacon, comminuted cured ham, and canned, shelf-stable luncheon meat.



Nitrosamines (亞硝胺)

- When nitrite reacts with secondary amines, **nitrosamines** are formed, and many are known to be **carcinogenic**.
- Tertiary amines and quaternary ammonium compounds also yield nitrosamines with nitrite under acidic conditions.



Nitrite-Sorbate and Other Nitrite Combinations

- In an effort to reduce the potential hazard of nitrosamine formation in bacon, the USDA in 1978 reduced the input NO2 level for bacon to 120 ppm and set a 10-ppb maximum level for nitrosamines.
- A proposal to allow the use of 40 ppm nitrite in combination with 0.26% potassium sorbate for bacon was made in 1978 but rescinded a year later when taste panel studies revealed undesirable effects.
- Many studies have shown that 0.26% sorbate in combination with 40 or 80 ppm nitrite is effective in preventing botulinal toxin production. (Table 13-2)



Nitrite-Sorbate and Other Nitrite Combinations

- EDTA at 500 ppm and chelate, 8hydroxyquinoline at 200 ppm have been evaluated as a nitrite-sparing agent.
- Potassium sorbate significantly decreased toxin production by types A and B spores in pork slurries when NaCl was increased or pH and storage temperature were reduced.



Mode of Action

- Nitrite inhibits *C. botulinum* by reacting with **iron-sulfur enzymes** such as ferredoxin and thus **preventing the synthesis of ATP from pyruvate**.
 - The phosphoroclastic system of *C*. *sporogenes* and *C*. *botulinum* is inhibited by nitric oxide → accumulation of pyruvic acid in the medium.
 - The phosphoroclastic reaction involves the breakdown of pyruvate with inorganic phosphate and coenzyme A to yield acetyl phosphate. In the presence of ADP, ATP is synthesized from acetyl phosphate with acetate as the other product.





Mode of Action

- Nitric oxide reacted with iron—sulfur complexes to form iron-nitrosyl complexes.
 → destruction of iron-sulfur enzymes such as ferredoxin.
- The resistance of the lactic acid bacteria to nitrite inhibition ←lack ferredoxin



Summary of Nitrite Effects

- When added to processed meats, nitrite has definite antibotulinal effects. It also forms desirable product color and enhances flavor in cured meat products.
- The antibotulinal effect:
 - inhibition of vegetative cell growth and
 - the prevention of germination and growth of spores



Summary of Nitrite Effects

- Clostridia other than *C. botulinum* are affected in a similar manner. Whereas low initial levels of nitrite are adequate for color and flavor development, considerably higher levels are necessary for the antimicrobial effects.
- When nitrite is heated in certain laboratory media→ produce Perigo effect/factor or Perigo inhibitor. It does not form in filtersterilized media.
 - It develops in canned meats only when nitrite is present during heating. Once formed, the Perigo factor is not affected greatly by pH changes.



Summary of Nitrite Effects

- Measurable levels of nitrite decrease considerably during heating in meats and during postprocessing storage—more at higher storage temperatures than at lower.
- The antibotulinal activity of nitrite is interdependent with pH, salt content, temperature of incubation, and numbers of botulinal spores. Heat-injured spores are more susceptible to inhibition than uninjured.



Summary of Nitrite Effects

- Lactic acid bacteria are relatively resistant to nitrite.
- **Endospores remain viable in the presence of the antibotulinal effect** and will germinate when transferred to nitrite-free media.
- Nitrite has a pK of 3.29 and, consequently, exists as undissociated nitrous acid at low pH values. The maximum undissociated state and consequent greatest antibacterial activity of nitrous acid are between pH 4.5 and 5.5.



NaCl and Sugars

- At high concentrations, salt exerts a drying effect on both food and microorganisms.
- **0.85-0.90%** salt produces an **isotonic** condition for nonmarine microorganisms.
- When microbial cells are suspended in a 5% saline solution→ the cell is plasmolysis→ growth inhibition and possibly death.
- When high concentrations of salt are added to fresh meats
 → both the microbial cells and those of the meat undergo plasmolysis (shrinkage), → drying of the meat + inhibition or death of microbial cells. ← Enough salt must be used to effect hypertonic conditions.
- The inhibitory effects of salt are **not dependent on pH**. Most nonmarine bacteria can be inhibited by **20% or less of NaCl**, whereas some molds generally tolerate higher levels.



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NaCl and Sugars

- Organisms that can grow in the presence of and **require high concentrations of salt** are referred to as **halophiles;** those that can withstand but not grow in high concentrations are referred to as **halodurics.**
- Sugars, such as sucrose, exert their preserving effect in essentially the same manner as salt. One of the main differences is in **relative concentrations**.
 - It generally requires about six times more sucrose than NaCl to effect the same degree of inhibition.
- The most common uses of sugars as preserving agents are in the making of fruit preserves, candies, condensed milk, and the like. → high concentrations of sugar makes water unavailable to microorganisms.



Indirect Antimicrobials

- The compounds and products in this section are multifunctional food additives.
- They are added to foods primarily for effects other than antimicrobial.
 - Antioxidants
 - Flavoring agents (including spices and essential oils)
 - Phosphates
 - Medium-chain fatty acids and esters



NaCl and Sugars

- Microorganisms differ in their response to hypertonic concentrations of sugars, with yeasts and molds being less susceptible than bacteria.
 - Some yeasts and molds can grow in the presence of as much as 60% sucrose, whereas much lower levels inhibit most bacteria.
- Organisms that are able to grow in high concentrations of sugars are designated osmophiles (嗜高溶微生物); osmoduric microorganisms are those that are unable to grow but are able to withstand high levels of sugars.



Antioxidants

- Although used in foods primarily to prevent the autooxidation of lipids, many phenol antioxidants (Table 13-8) have been shown to possess antimicrobial activity against a wide range of microorganisms.
- These compounds have been evaluated extensively as **nitrite-sparing agents** in processed meats and in combination with other inhibitors.



Flavoring Agents

- Impart aromas and flavors to foods
- more **antifungal** than antibacterial.
- The nonlactic, Gram-positive bacteria are the most sensitive, and the lactic acid bacteria are rather resistant.
- The **essential oils** and **spices** have received the most attention by food microbiologists
- One of the most effective flavoring agents is **diacetyl**, which imparts the **aroma of butter**. It is more effective against Gramnegative bacteria and fungi than against Gram-positive bacteria.



Flavoring Agents

- Diacetyl inhibits **arginine utilization** by reacting with arginine-binding proteins of Gram-negative bacteria.
- Many spices possess significant antimicrobial activity. Their antimicrobial activities are due to specific chemicals or essential oils.
- At least 20 spices or their extracts against most food-poisoning organisms, including mycotoxigenic fungi.



Phosphates

- Added to processed meats to increase their water holding capacity.
- Food-grade phosphates range from one P (e.g. trisodium phosphate, TSP) to at least 13 (sodium polyphosphate).
- They possess antibotulinal activity, especially when combined with nitrites.
- Filter-sterilized phosphate preparations were more inhibitory than autoclaved in one study.
- Active growth of cells was necessary for its bactericidal effect.



Medium-Chain Fatty Acids and Esters

- Acetic, propionic, and sorbic acids are short-chain fatty acids used primarily as preservatives.
- Medium-chain fatty acids are employed primarily as surface-active or emulsifying agents.
- The antimicrobial activity of the medium-chain fatty acids is best known from **soaps**, which are salts of fatty acids. Those most commonly employed are composed of **12-16 carbons**.



Medium-Chain Fatty Acids and Esters

- In general, fatty acids are effective primarily against gram-positive bacteria and yeasts.
- The monoesters of glycerol and the diesters of sucrose are more antimicrobial than the corresponding free fatty acids and better than sorbic acid and the parabens as antimicrobials.
- Monolaurin is the most effective of the glycerol monoesters, and sucrose dicaprylate is the most effective of the sucrose diesters.



Preservative system

- Using combinations of chemicals
- Consist of three compounds monolaurin/EDTA/BHA, for example.
- Although EDTA possesses little antimicrobial activity by itself, it makes gram-negative bacteria more susceptible by rupturing the outer membrane and thus enhances the effect of fatty acids or fatty acid esters.
- Antioxidant BHA would exert effects against bacteria and molds and also serve as an antioxidant
- By use of such a system, the development of resistant strains could be minimized and the pH of a food could become less important relative to the effectiveness of the inhibitory system.



Acetic and Lactic Acids

- The antimicrobial effects of organic acids such as propionic and lactic is due to both the depression of pH below the growth range and metabolic inhibition by the undissociated acid molecules.
- In determining the quantity of organic acids in foods, **titratable acidity** is of more value than pH alone, because the latter is a measure of hydrogen-ion concentration and **organic acids do not ionize completely**.



Acetic and Lactic Acids

- In measuring titratable acidity, the amount of acid that is capable of reacting with a known amount of base is determined. → ? volume of 0.1N alkali reacted/100g or 100 ml of original material
- The bactericidal effect of acetic acid : When two species of *Salmonella* were added to an oiland-vinegar-based salad dressing, the initial inoculum of 5 x 10⁶ *S. enteritidis* could not be detected after 5 min nor could *S. typhimurium* be detected after 10 min.
- Organic acids are employed to wash and sanitize animal carcasses after slaughter to reduce their carriage of pathogens and to increase product shelf life.



Antibiotics

- Antibiotics are secondary metabolites (which are not formed during the exponential growth phase and have no apparent significance to the growth or metabolism) produced by microorganisms that inhibit or kill a wide spectrum of other microorganisms.
- Most antibiotics are produced by molds and bacteria of the genus *Streptomyces*, and a few by *Bacillus* and *Paenibacillus* spp.
- Many of the clinically useful agents are synthetic products.



Antibiotics (Table 13-9)

- The general view in USA is that the benefits to be gained by using antibiotics in foods do not outweigh the risks.
- Several key considerations on the use of antibiotics as food preservatives are summarized as follows:
 - The antibiotic agent should kill, not inhibit, the flora and should ideally decompose into innocuous products or be destroyed on cooking for products that require cooking.
 - The antibiotic **should not be inactivated** by food components or products of microbial metabolism.
 - The antibiotic should not readily stimulate the appearance of resistant strains.
 - The antibiotic should not be used in foods if used therapeutically or as an animal feed additive.



Monensin (孟寧素)

- Approved by FDA as a cattle additive in the 1970, and it is used primarily to improve feed efficiency in ruminants.
- Inhibits Gram-positive bacteria
- Like nisin, monensin is an ionophore (離子 載體, a lipid-soluble molecule that transports ions across a cell membrance)
 → destroys selective permeability of cell membranes.



Natamycin (納他黴素)

- Used as a **food preservative** with the consideration of the following facts:
- It does not affect bacteria,
- it stimulates an unusually low level of resistance among fungi,
- it is **rarely involved in cross-resistance** among other antifungal polyenes,
- DNA transfer between fungi is less than bacteria.



Natamycin

- Quite effective against yeasts and molds but not bacteria.
- Isolated from *Streptomyces natalensis*.
- Natamycin appears to act in the same manner as other polyene antibiotics—by binding to membrane sterols and inducing distortion of selective membrane permeability.
 - Because bacteria do not possess membrane sterols, their lack of sensitivity to this agent is thus explained.



Tetracyclines (四環黴素)

- Chlortetracycline (CTC) and oxytetracycline (OTC) delay bacterial spoilage of seafoods, poultry, red meats, vegetables, raw milk, and other foods.
- CTC is generally more effective than OTC.
- The tetracyclines are both **heat sensitive** and **storage labile** in foods.
- They are used to treat diseases in humans and animals and are used also in feed supplements.



Subtilin (枯草桿菌素)

- Structurally similar to nisin (Fig. 13-6)
- Produced by *Bacillus subtilis*.
- Like nisin, it is effective against grampositive bacteria, is stable to acid, and possesses enough heat resistance to withstand destruction at 121 °C for 30-60 min.
- Effective in canned foods at levels of 5-20 ppm in preventing the outgrowth of germinating endospores



Tylosin (泰黴素)

- More inhibitory than nisin or subtilin.
- Used in animal feeds and also to treat some diseases of poultry
- Efective against gram-positive bacteria
- Inhibits protein synthesis by associating with the 50S ribosomal subunit.



Antifungal Agents for Fruits

- Table 13-10 (thiabendazole, benomyl, biphenyl, SO₂ fumigation)
- **Benomyl** is applied uniformly over the entire surface of fruits. It can penetrate the surface of some vegetables and is used worldwide to control crown rot and anthracnose of bananas, and stem-end rots of citrus fruits.



Ethylene and Propylene Oxides (氧化乙烯及氧化丙烯)

- Exist as gases and are employed as fumigants in the food industry.
- Applied to dried fruits, nuts, spices, and so forth, primarily as **antifungal compounds.**
- are **alkylating agent**.



Ethylene oxide

- Its antimicrobial activity is presumed to be related to **alkylation**.
- In the presence of labile H atoms, the unstable threemembered ring of ethylene oxide splits.
- The H atom attaches itself to the oxygen, forming a hydroxyl ethyl radical, •CH2CH2OH
- •CH2CH2OH attaches itself to the position in the organic molecule left vacant by the H atom. The hydroxyl ethyl group blocks reactive groups within microbial proteins, thus resulting in inhibition.
- Among the groups capable of supplying a labile H atom are —COOH, —NH₃, —SH, and — OH.



Ethylene oxide

- Affect endospores of *C. botulinum* by alkylation of guanine and adenine components of spore DNA .
- Used as a **gaseous sterilant** for flexible and semirigid containers for packaging aseptically processed foods.
- Similarly effective against vegetative cells and endospores.



BIOCONTROL

- The use of one or more organisms to inhibit or control other organisms.
- May require a living organism (such as phages) or it may be effected by indirect actions or agents (such as bacteriocins).
- Related to the food protection provided by the activities of the lactic acid bacteria, bacteriocins, endolysins (溶菌酶), bacteriophages, and "protective cultures" in general.



Microbial Interference

- The general **nonspecific inhibition or destruction** of one microorganism by other members of the same habitat or environment.
- The mechanisms are not clear, but some observations are worthy of note.
 - The background biota needs to be **larger in a number of viable cells** than the organism to be inhibited.
- The interfering biotas generally **not homogeneous**, and the specific roles that individual species play are unclear.



Microbial Interference

- Possible explanations:
 - Competition of nutrients
 - Competition for attachment/adhesion sites
 - Unfavorable alteration of the environment
 - Combinations of these.

Lactic antagonism (拮抗作用)

- Lactic acid bacterium inhibits or kills closely related and food-poisoning and food spoilage organisms. → The most effective method used was spraying the lactic organism on food surfaces
- Precise mechanisms are not clear. The possible factors identified are:
 - the production of antibiotics, H₂O₂, diacetyl, and bacteriocins
 - pH depression
 - nutrient depletion



Lactic antagonism

- Protective cultures: the microorganisms that can be found in or added to a food product to effect preservation/protection. They should
 - present no health risks
 - provide beneficial effects on the product
 - have no negative impact on sensory properties
 - serve as "indicators" under abuse conditions

Nisin

- Nisin is a **bacteriocin**.
- Like antibiotics, bacteriocins are chemical compounds produced by microorganisms that inhibit or other microorganisms.
- Bacteriocins inhibit or kill generally only closely related species or strains of the same species.

Nisin (乳酸鏈球菌素)

- A polypeptide → structurally related to subtilin (Fig. 13-6).
- the most widely used antibiotic for food preservation
- **desirable properties as a food preservative** are the following:
 - a) nontoxic.
 - b) produced naturally by *Lactococcus lactis* strains.
 - c) heat stable and has excellent storage stability.
 - d) destroyed by digestive enzymes.
- e) does not contribute to off-flavors or off-odors.
- f) has a narrow spectrum of antimicrobial activity.



Nisin

- Effective against **Gram-positive** bacteria, primarily **sporeformers**, and ineffective against fungi and gram-negative bacteria.
- Mode of action
 - nisin and subtilin appear to be identical.
- react with cytoplasmic membranes and result in pore formation.
- The formation of pore causes the loss of accumulated amino acids and the inhibition of amino acid transport.



ENDOLYSINS

- In order to release newly formed bacteriophages from their host cell, two small hydrophobic proteins were used:
- Holins disrupt the cell membrane and form holes through which endolysins can pass.
- **Endolysins** target bonds in the peptidoglycan, and upon the destruction of this cell barrier, the phage progeny is released.



ENDOLYSINS

- In addition to their lysis of bacterial cells from within, endolysins from Gram-positive bacteria phages also lyse bacteria exogenously.
- Phage endolysin can control some foodborne bacterial pathogens.



BACTERIOPHAGES AS BIOCONTROL AGENTS

- Lytic phages can destroy their specific host cells in foods.
- The true potential need more research

THE HURDLE CONCEPT

- Multiple factors or techniques are employed to effect the control of microorganisms in foods.
- Barrier technology, combination preservation, and combined methods

