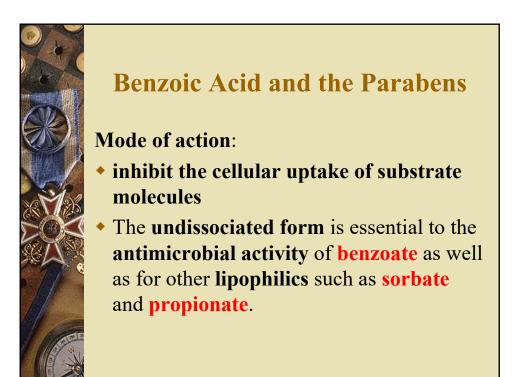


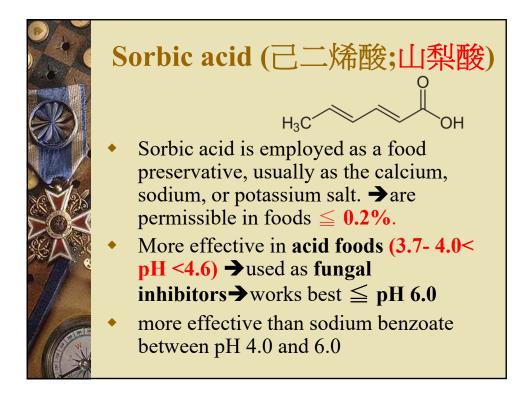


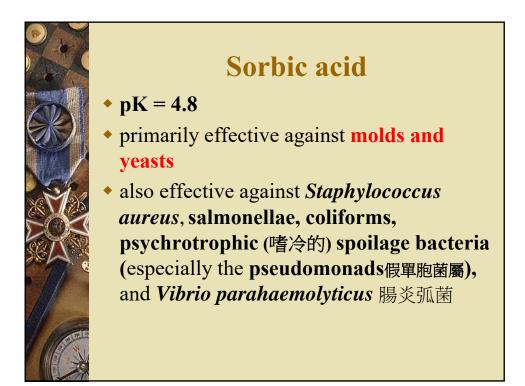


Benzoic Acid and the Parabens

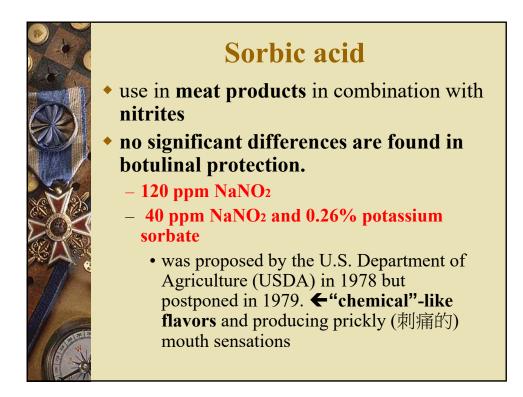
- antimicrobial activity → undissociated
 molecule → most active at the lowest pH
- The pK of benzoate is 4.2 → used in high-acid products (pH<4.0-3.7) → as a mold and yeast inhibitor
- The pK for the parabens is around 8.47.
 → effective at pH ≤ 8.0→ as a mold and yeast inhibitor

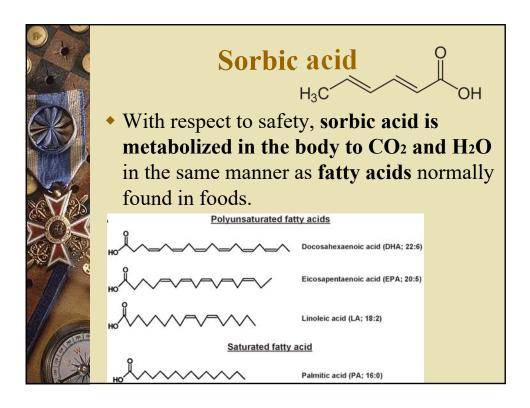


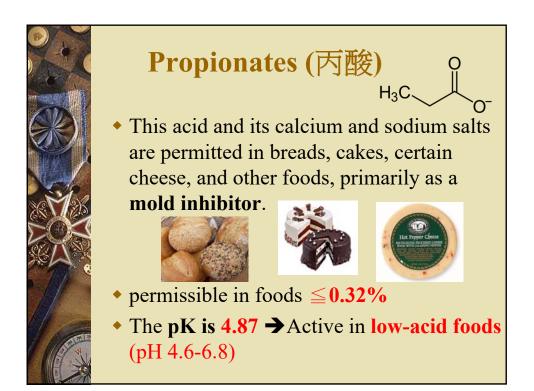














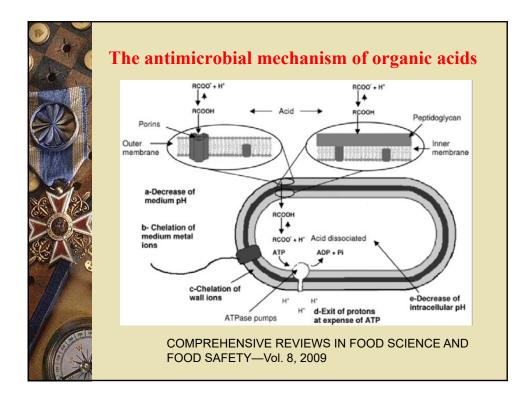
The antimicrobial mechanism of **lipophilic** acids (sorbate, benzoate, and propionate)

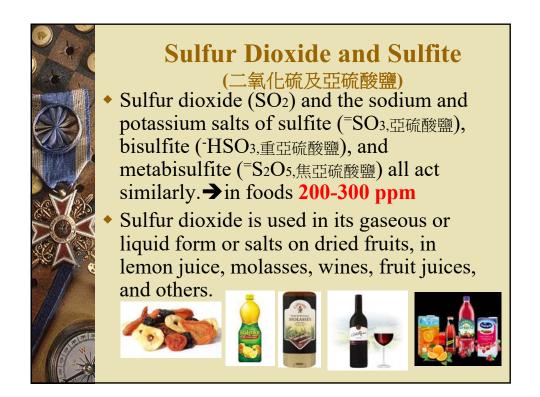
- Involves the proton motive force (PMF 質子驅動力)
- Hydrogen ions (protons) and hydroxyl ions are separated by the cytoplasmic membrane, hydrogen ions (outside the cell) giving rise to acidic pH and hydroxyl ions (inside the cell) giving rise to pH near neutrality.



The antimicrobial mechanism of **lipophilic**

- acids (sorbate, benzoate, and propionate)
- The membrane gradient represents electrochemical potential that the cell employs in the active transport of some compounds such as amino acids.
- After these weak lipophilic acids diffusing across the membrane, the undissociated molecule ionizes inside the cell and lowers intracellular pH. → a weakening of the transmembrane gradient such that amino acid transport is affected adversely

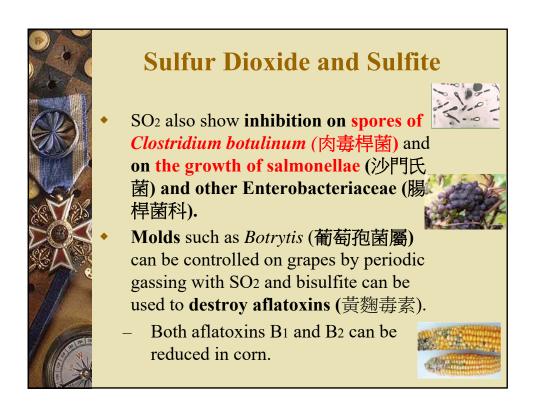






Sulfur Dioxide and Sulfite

- Because sulfites destroy thiamin, they are not
 permitted in meats or other foods
 recognizable as sources of thiamine (Vitamin B1).
- The sulfites react with various food constituents including **nucleotides**, **sugars**, **disulfide bonds**, and others.
- Also used as an **antioxidant**.
- SO₂ is **bacteriostatic** against *Acetobacter* **spp**. (醋酸桿菌屬) and the **lactic acid bacteria** at low pH, concentrations of 100-200 ppm being effective in fruit juices and beverages. It is **bactericidal at higher concentrations**.





Sulfur Dioxide and Sulfite

- The actual mechanism of action of SO_2 is not known.
- One suggestion is that the undissociated sulfurous acid (亞硫酸 H₂SO₃) or molecular SO₂ is responsible for the antimicrobial activity (Its greater effectiveness at low pH tends to support this).
- The other suggestion is that the antimicrobial action is due to the strong reducing power that allows these compounds to reduce oxygen tension to a point below that at which aerobic organisms can grow or by direct action on some enzyme system.

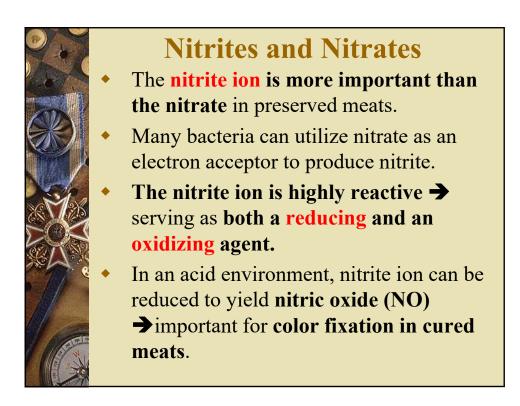


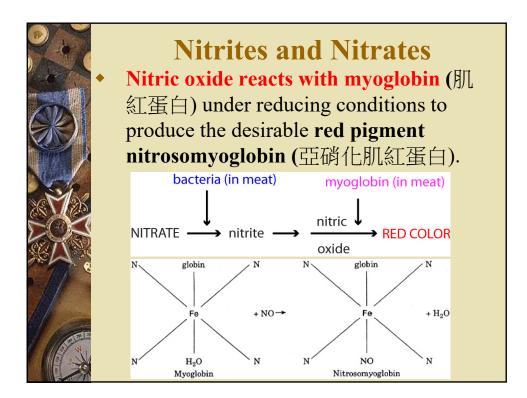
Sulfur Dioxide and Sulfite

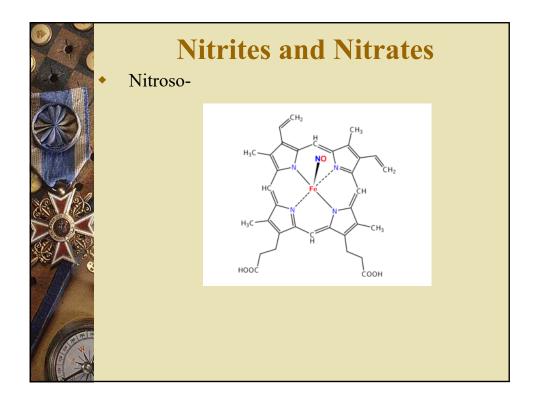
- SO₂ is also thought to be an enzyme poison, inhibiting growth of microorganisms by inhibiting essential enzymes.
 - Its use in the drying of foods to inhibit enzymatic browning is based on this assumption.
 - Because the sulfites are known to **act on disulfide bonds**, it may be presumed that certain essential enzymes are affected.

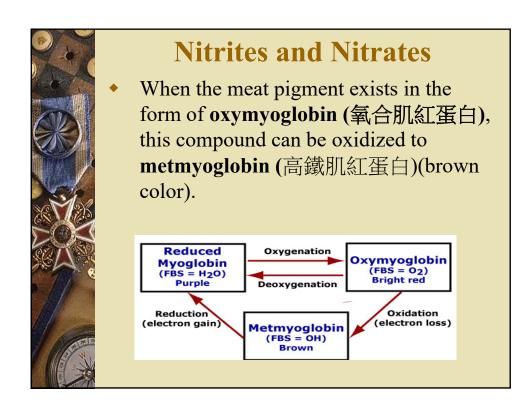


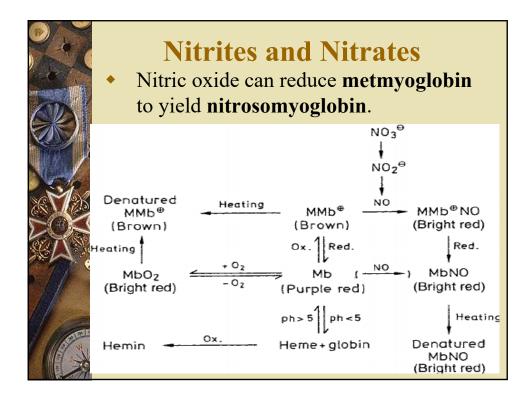


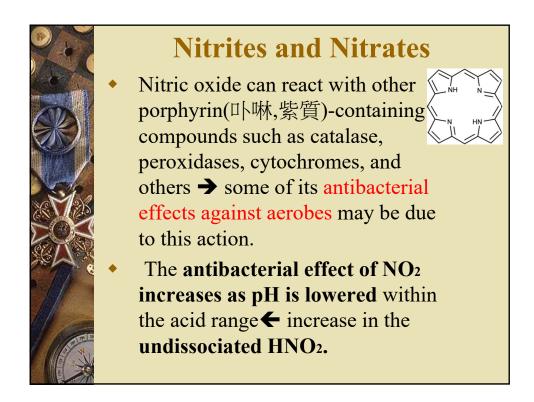


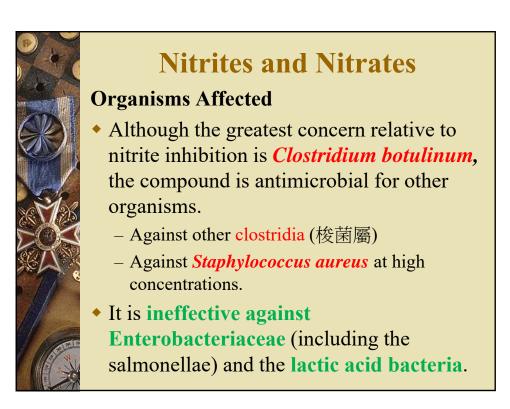


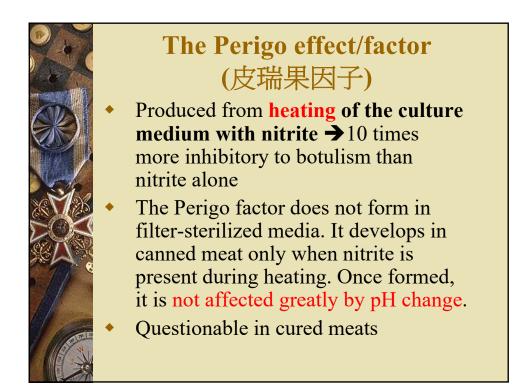




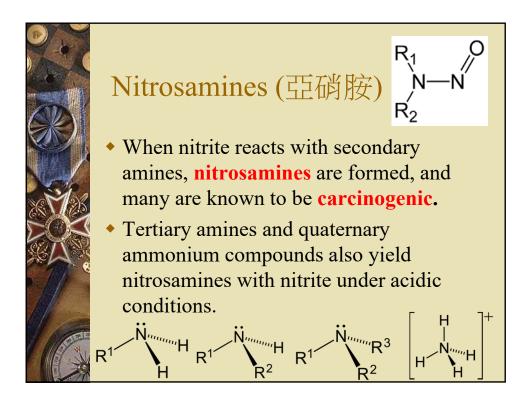










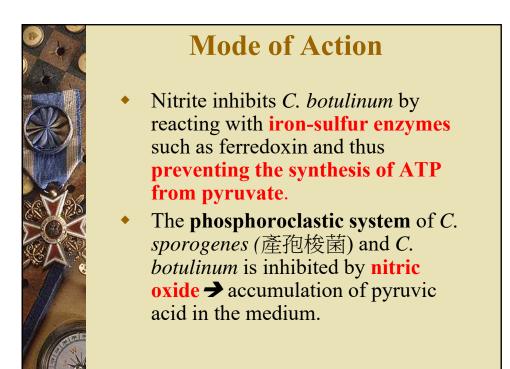




Nitrite-Sorbate and Other Nitrite Combinations

- In an effort to reduce the potential hazard of nitrosamine formation in bacon, the USDA in 1978 reduced the input NO2 level for bacon to **120 ppm** and set a **10ppb** maximum level for nitrosamines.
 - A proposal to allow the use of **40 ppm nitrite in combination with 0.26% potassium sorbate** for bacon was made in 1978 but rescinded a year later when taste panel studies revealed undesirable effects.







Mode of Action

The phosphoroclastic reaction involves the breakdown of **pyruvate** with **inorganic phosphate and coenzyme A** to yield **acetyl phosphate**. In the presence of ADP, ATP is synthesized from acetyl phosphate with **acetate** as the other product.

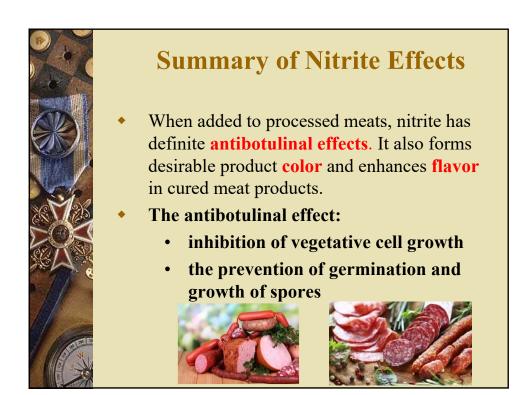


Electrons are transferred first to **ferredoxin** and from ferredoxin to H^+ to form H_2 in a reaction catalyzed by **hydrogenase**. Ferredoxin and **hydrogenase** are iron-sulfur (nonheme) proteins or enzymes.



Mode of Action

- Nitric oxide reacted with iron—sulfur complexes to form iron-nitrosyl complexes.
 → destruction of iron-sulfur enzymes such as ferredoxin.





Summary of Nitrite Effects

Clostridia other than *C. botulinum* are affected in a similar manner. Whereas **low initial levels** of nitrite are adequate for **color and flavor** development, considerably **higher levels are necessary for the antimicrobial effects**.

When nitrite is heated in media or canned meats \rightarrow produce Perigo effect/factor or Perigo inhibitor. It is not affected greatly by pH changes.



Summary of Nitrite Effects

- Measurable levels of nitrite decrease considerably during heating in meats and during postprocessing storage—more at higher storage temperatures than at lower.
- The antibotulinal activity of nitrite is interdependent with pH, salt content, temperature of incubation, and numbers of botulinal spores. Heat-injured spores are more susceptible to inhibition than uninjured.



Summary of Nitrite Effects

- Lactic acid bacteria are relatively resistant to nitrite.
- **Endospores remain viable in the presence of the antibotulinal effect** and will germinate when transferred to nitrite-free media.
- Nitrite has a pK of 3.29 and exists as undissociated nitrous acid at low pH values. The maximum undissociated state and consequent greatest antibacterial activity of nitrous acid (nitrite) are between pH 4.5 and 5.5.



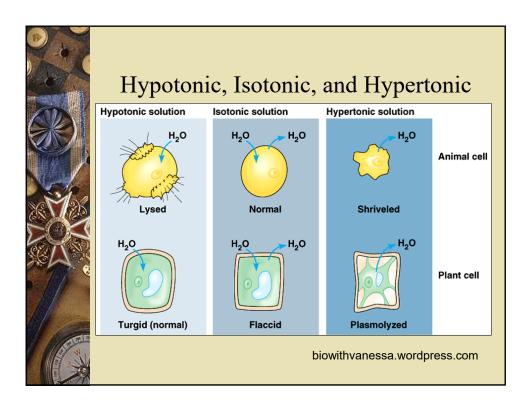
NaCl and Sugars

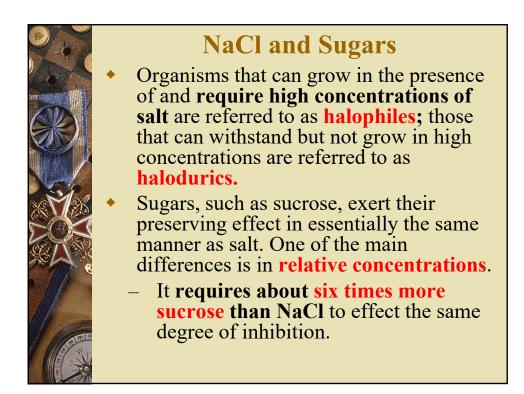
- At high concentrations, salt exerts a **drying effect** on both food and microorganisms.
- 0.85-0.90% salt produces an isotonic condition for nonmarine microorganisms.
- When microbial cells are suspended in a 5% saline solution → the cell is plasmolysis → growth inhibition and possibly death.



NaCl and Sugars

- When high concentrations of salt are added to fresh meats → both the microbial cells and those of the meat undergo plasmolysis (shrinkage)→drying of the meat + inhibition or death of microbial cells ← Enough salt must be used to effect hypertonic conditions.
- The inhibitory effects of salt are not dependent on pH. Most nonmarine bacteria can be inhibited by 20% or less of NaCl, whereas some molds generally tolerate higher levels.



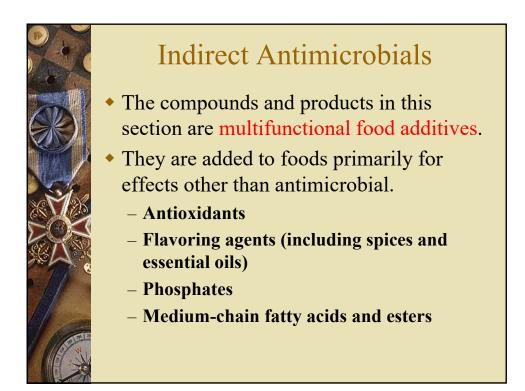






NaCl and Sugars

- Microorganisms differ in their response to hypertonic concentrations of sugars, with yeasts and molds being less susceptible than bacteria.
 - Some yeasts and molds can grow in 60% sucrose, whereas much lower levels inhibit most bacteria.
- Organisms that are able to grow in high concentrations of sugars are designated osmophiles (嗜高滲透壓生物); osmoduric (耐高滲透壓) microorganisms are those that are unable to grow but are able to withstand high levels of sugars.





Antioxidants

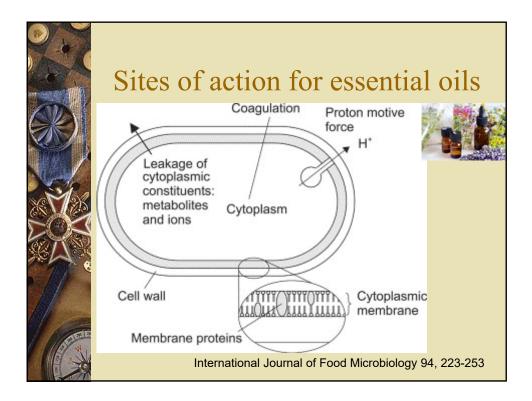
- Although used in foods primarily to prevent the autooxidation of lipids, many phenol antioxidants (Table 13-8) have been shown to possess antimicrobial activity against a wide range of microorganisms.
- These compounds have been evaluated extensively as **nitrite-sparing agents** in processed meats and in combination with other inhibitors.

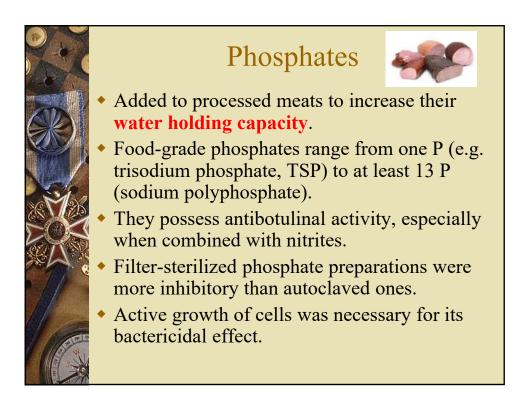


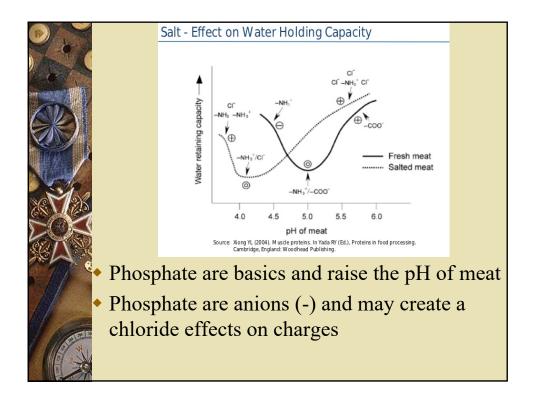


Flavoring Agents

- Diacetyl inhibits arginine utilization by reacting with arginine-binding proteins of Gram-negative bacteria.
- Many spices possess significant antimicrobial activity. Their antimicrobial activities are due to **specific chemicals** or **essential oils**.



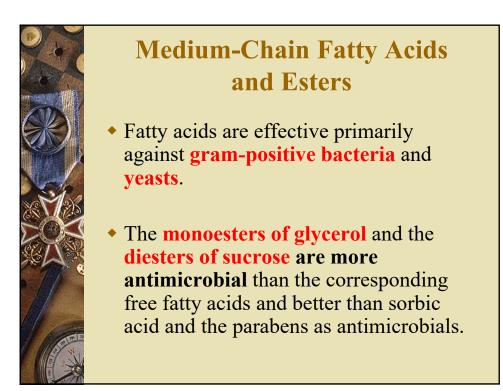


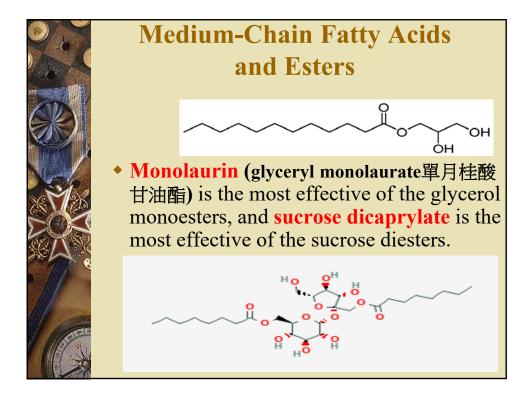


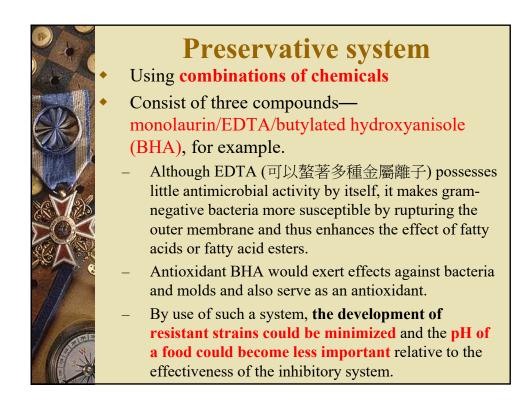


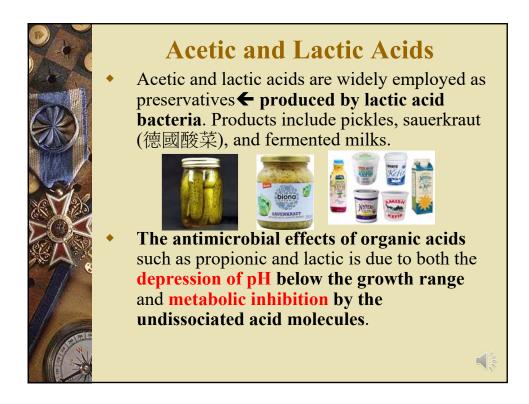
Medium-Chain Fatty Acids and Esters

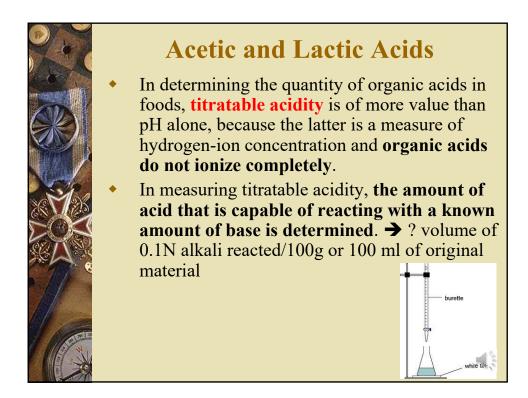
- Acetic, propionic, and sorbic acids are **short-chain fatty acids** used primarily as preservatives.
- Medium-chain fatty acids are employed primarily as surface-active or emulsifying agents.
- The antimicrobial activity of the medium-chain fatty acids is best known from soaps, which are salts of fatty acids. Those most commonly employed are composed of 12-16 carbons.

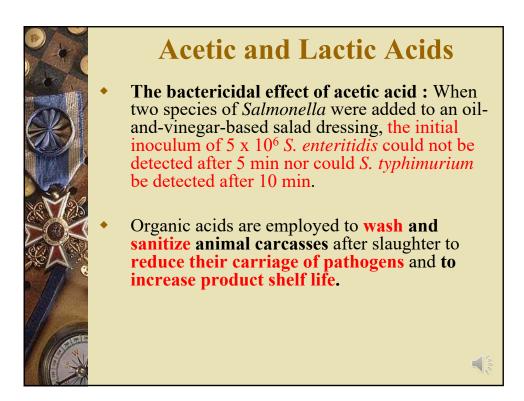


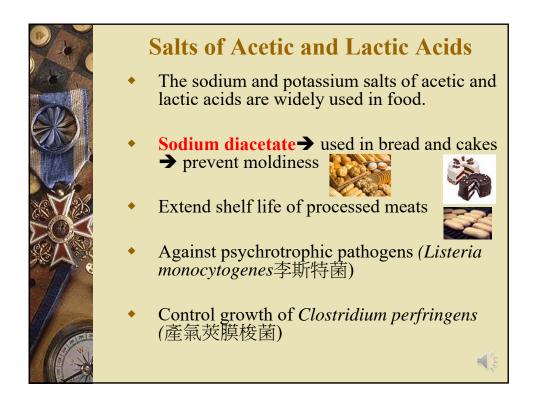








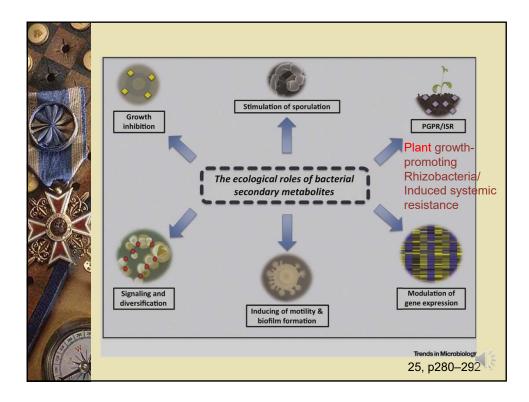


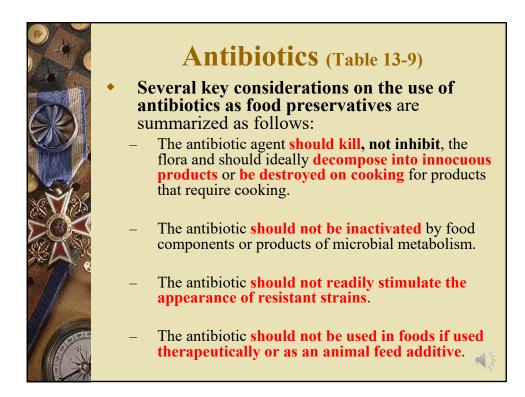


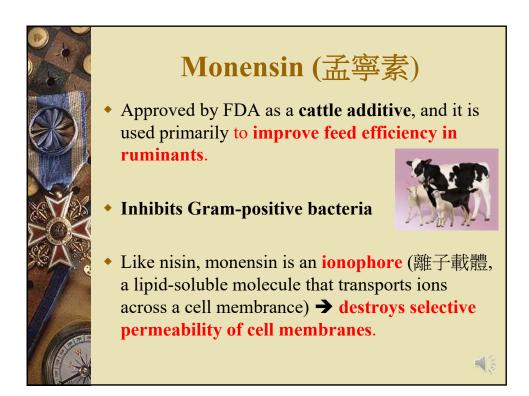


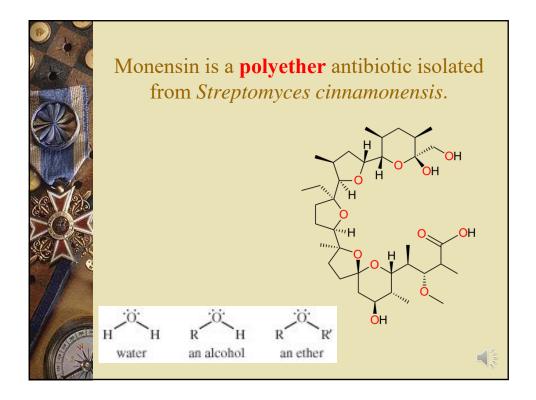
Antibiotics

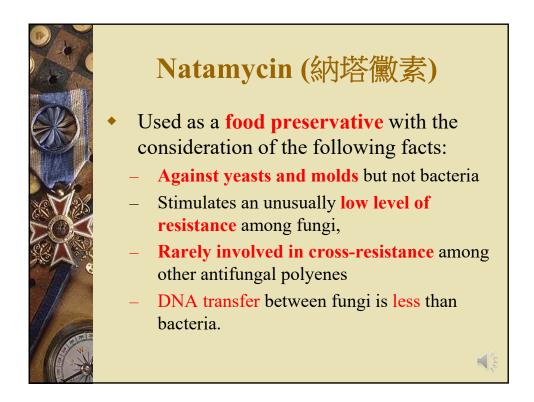
- Antibiotics are secondary metabolites produced by microorganisms that inhibit or kill
- a wide spectrum of other microorganisms.
 - Primary metabolites are considered essential to microorganisms for proper growth.
- Secondary metabolites have no apparent significance to the growth or metabolism and are usually formed during the stationary phase of growth.
- Most antibiotics are produced by molds and bacteria of the genus Streptomyces (鏈絲菌屬), and a few by Bacillus (芽孢桿菌屬) and Paenibacillus (類芽孢桿菌屬).
- Many of the clinically useful agents are synthetic products.

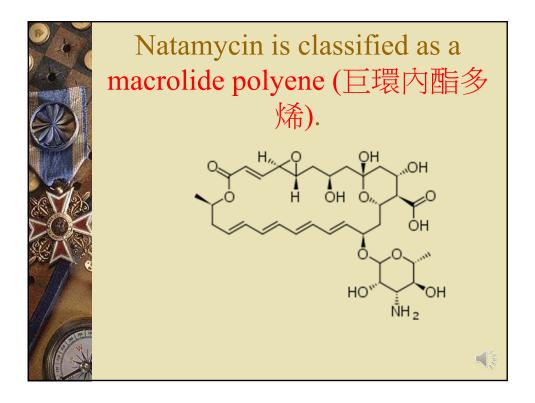


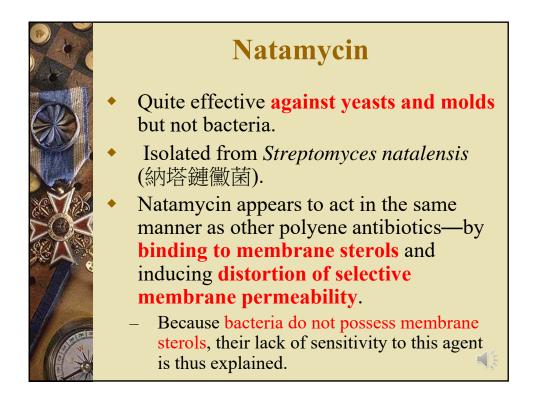


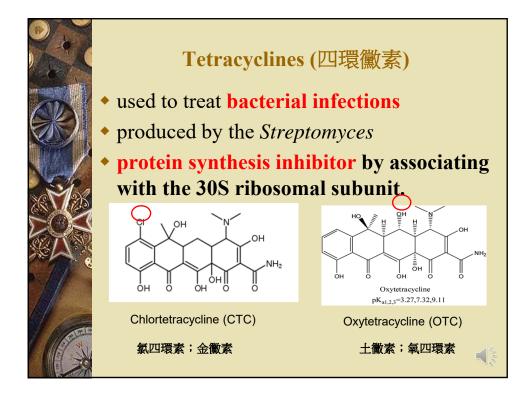


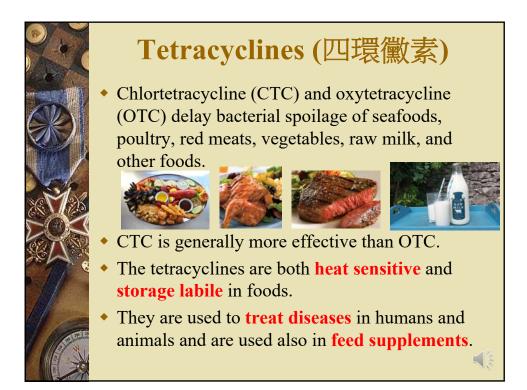


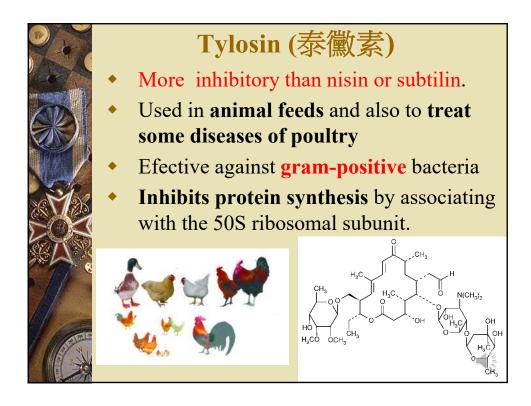














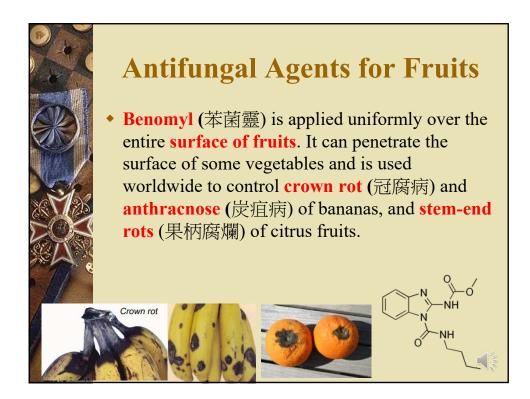
Antifungal Agents for Fruits

Table 13–10 Some Chemical Agents Employed to Control Fungal Spoilage of Fresh Fruits Compound Fruits Thiabendazole Apples, pears, citrus fruits, pineapples **Benomyl** Apples, pears, bananas, citrus fruits, mangoes, papayas, peaches, cherries, pineapples **Biphenyl** Citrus fruits

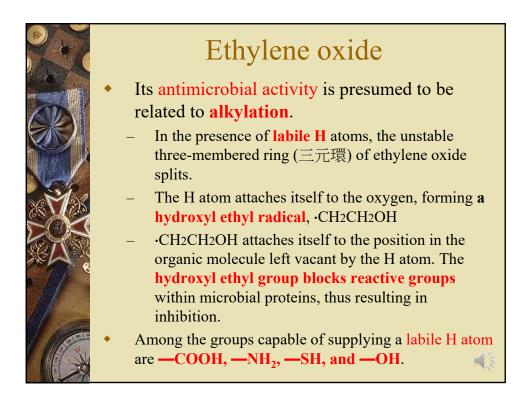
SO₂ fumigation Sodium-α-phenylphenate fruits, pineapples

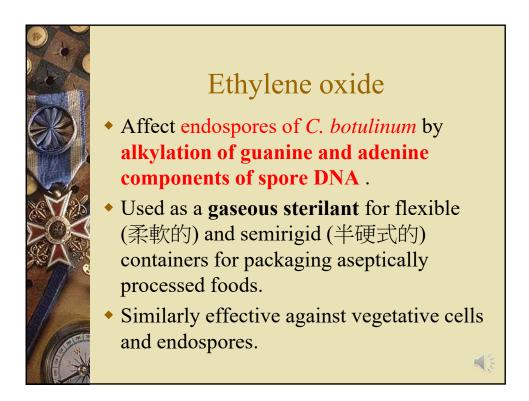
Grapes

Apples, pears, citrus





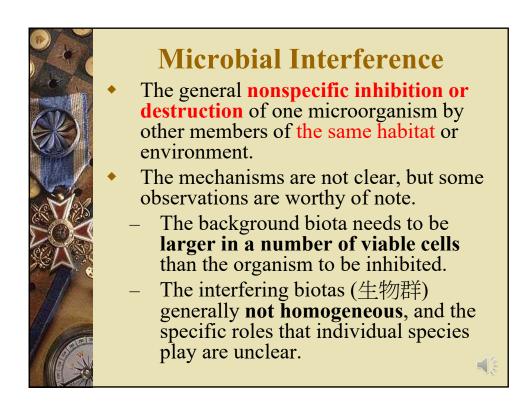


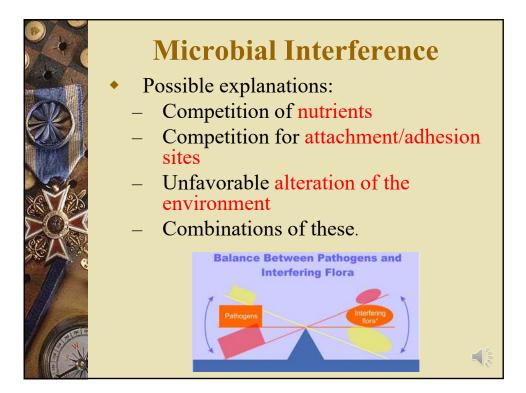


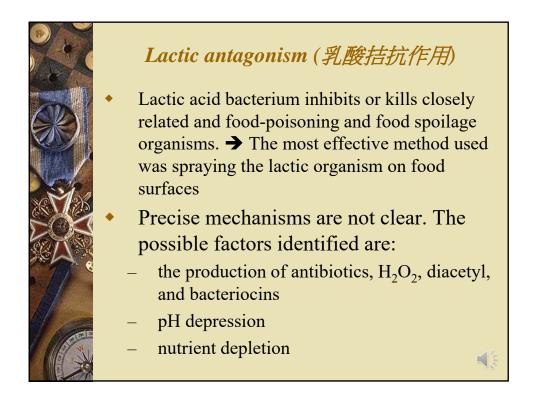


BIOCONTROL

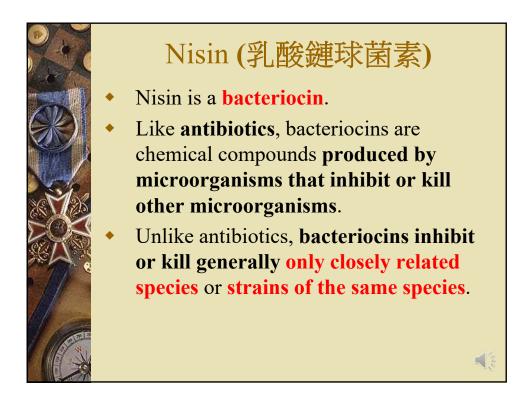
- The use of one or more organisms to inhibit or control other organisms.
- May require a living organism (such as lactic acid bacteria) or it may be effected by indirect actions or agents (such as the production of bacteriocins).
- Related to the food protection provided by the activities of the lactic acid bacteria, bacteriocins, endolysins (溶菌 酶), bacteriophages, and "protective cultures" in general.

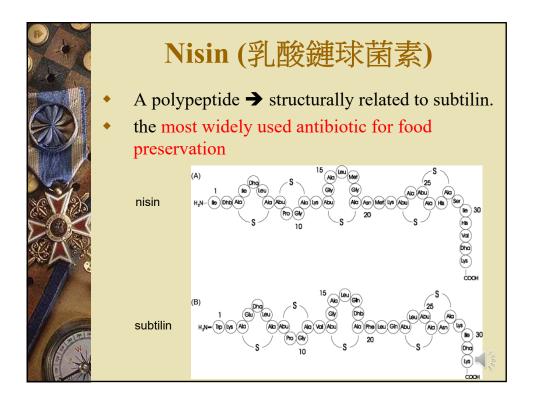






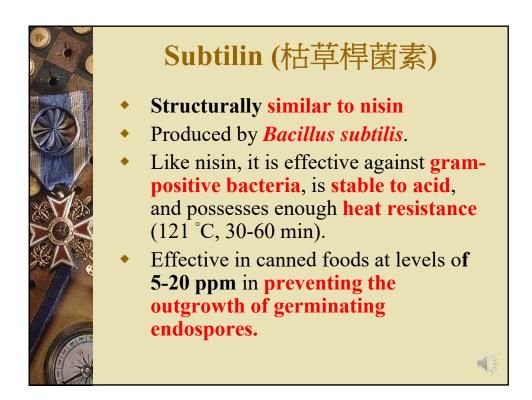














ENDOLYSINS (內溶素)

- **To release newly formed bacteriophages** from their host cell, two small hydrophobic proteins were used:
- Holins (穿孔素) disrupt the cell membrane and form holes through which endolysins can pass.
- Endolysins target bonds in the peptidoglycan, and upon the destruction of this cell
 barrier, the phage progeny is released.

